



Nano Environmental Technology
The Reliable Alternative



USER MANUAL & CHARACTERISATION DATA

NT-H2S-LI Electrochemical Hydrogen Sulphide Gas Sensor



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INTRODUCTION

Nemoto & Co. Ltd was established in 1941 and continues to develop unique technologies for the Safety, Security and Health markets. Using our unique experience of fine chemical preparation and printing, we were able to enter the gas sensor market in 1979 with a range of high-quality hot-wire type sensors (pellistors). Nemoto is now one of the World's leading manufacturers of chemical sensors and has so far delivered over 30-million devices to the market.

As a result of three years' development at our Tokyo R&D centre, we released our first electrochemical gas sensors in 2000.

The NT-H2S-LI Gas Sensor is a new 3-Electrode Electrochemical cell designed for the detection and measurement of hydrogen sulphide in the range 0-100ppm. By adhering to industry standards for size and connection orientation, the NT-H2S-LI can be retrofitted easily to existing product designs.

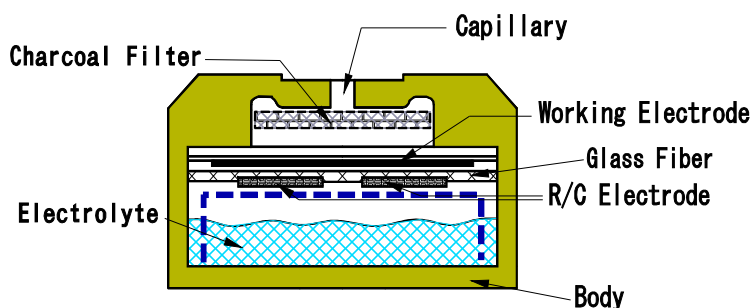
By using our experience of design for manufacture and our high volume production facilities in Japan and China, we have successfully reduced the cost of the NT-H2S-LI whilst being able to maintain the highest performance quality.

This characterisation document does not constitute a specification but is intended as a guide, informing the instrument designer of the performance characteristics of the sensor which were observed by Nemoto Environmental Technology's Engineers.

It should be read in conjunction with Technical Information Sheet DS-N-NT-H2S-LI, which includes the full technical specification for the NT-H2S-LI Gas Sensor.

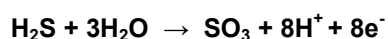


PRINCIPLES OF OPERATION



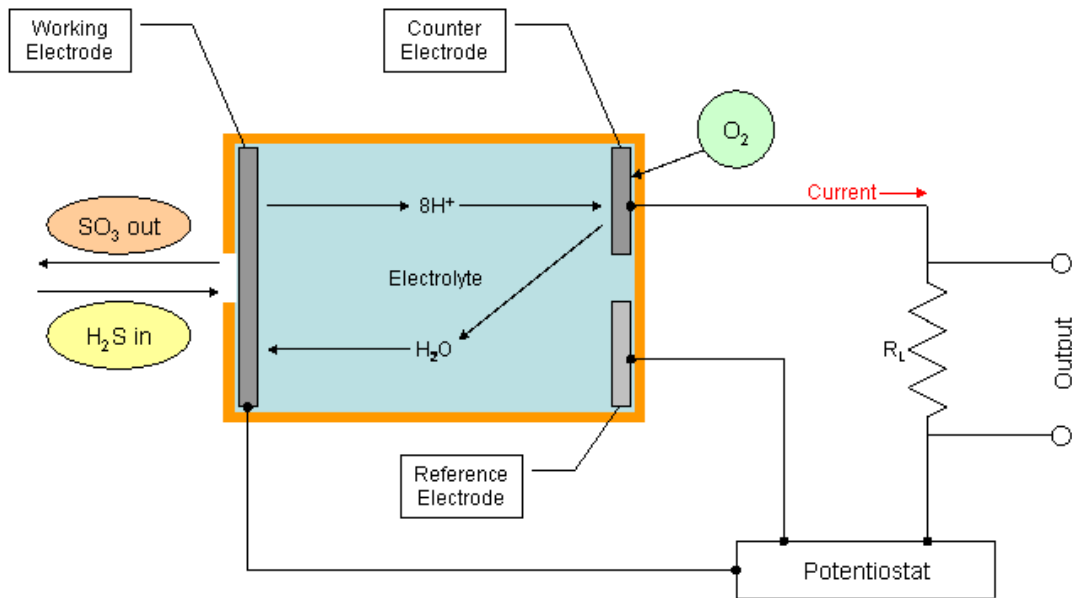
The NT-H₂S-LI consists of 3 porous noble metal electrodes separated by an acidic aqueous electrolyte, housed within a plastic (PPO) enclosure. Gas enters the cell via a gas phase diffusion barrier (capillary) and a charcoal based filter that removes unwanted gases which might interfere with the sensors performance or give a false signal. An electrolyte reservoir ensures an excess of electrolyte is available at all times, and the sensor is vented to ensure that the internal and external pressure of the sensor is in equilibrium.

In operation, gas enters the cell via the capillary and filter, and comes into contact with the 'working' electrode. Any hydrogen sulphide present undergoes the following (oxidation) reaction:



The SO₃ generated either vents away from the cell via the capillary or dissolves in water to form Sulphuric acid electrolyte, whilst the hydrogen ions (H⁺) migrate into the electrolyte within the cell. The electrons (e⁻) generated at the working electrode are collected by the external circuit via a metal strip in contact with it, in the form of a small (nA) electric current.

The reaction at the working electrode is balanced by a reciprocal (reduction) reaction at the 'counter' electrode, using Oxygen from the surrounding atmosphere. The electrons consumed in this reaction are supplied by the external circuit via a metal strip in contact with the counter electrode.



Thus water is consumed whilst Hydrogen ions are generated at the working electrode, whilst the water is re-created and hydrogen ions are consumed at the counter electrode. At the same time, the reaction at the working electrode generates electrons, whilst the reaction at the counter electrode consumes electrons. By connecting the working and counter electrodes together via a special circuit, flow of electrons between the two electrodes is measured as a nA level current signal proportional to the carbon monoxide concentration.

The 'reference' electrode maintains the healthy operation of the cell. It is surrounded by electrolyte, sees no gas and no current is allowed to be drawn from it. Its electrochemical potential hence always remains constant at a level known as the "rest air potential" and this is used to regulate the potential of the working electrode, regardless of the current it is generating during operation. The use of a reference electrode in this way (i.e. three-electrode operation) helps to extend the working range of the sensor, improves linearity and results in a number of performance benefits compared with similar sensors working with 2-electrodes only.



FEATURES

Electrochemical gas sensors have the following superiority to conventional semiconductor type and hot-wire type gas sensors.

- **Linear output in proportion to gas concentration**
- **High reproducibility**
- **Highly gas specific**
- **Unaffected by humidity**
- **Stable output for long periods**
- **Low power consumption because no heater is used (can be battery operated)**
- **Small and lightweight (can be used in portable devices)**
- **No mechanical structure so highly resistant to shocks and vibrations.**

NT-H2S-LI has been developed from our accumulation of technologies in production of hot-wire type gas sensors, long research experience into catalysts, fine printing, and assembling of sensors. The NT-H2S-LI is small and less-expensive, but has high sensitivity, long life, and leak-free performance even under severe operating conditions.

Air vent

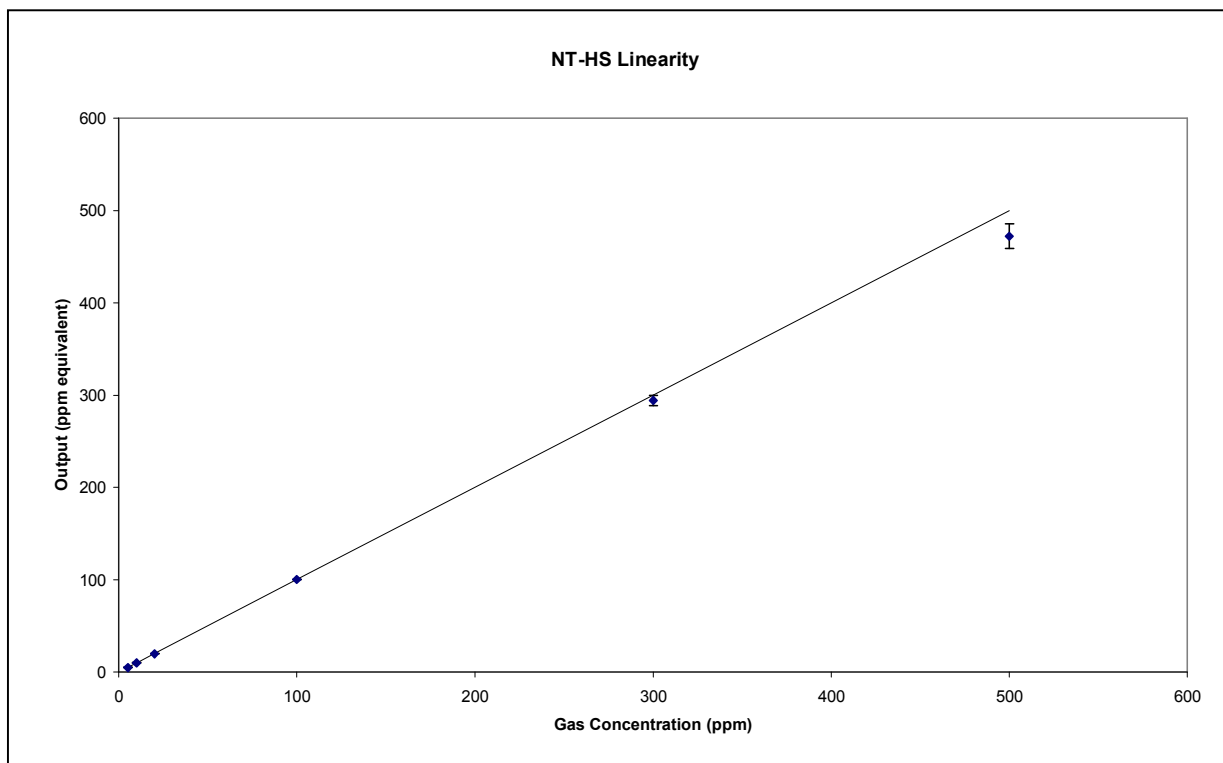
The electrolyte used for chemical sensors is very hygroscopic, i.e. it has affinity for water, and its volume varies depending on ambient temperature and humidity. This variation causes pressure inside the sensor to rise and fall. In the worst case the electrolyte may leak out of the sensor and damage the circuitry around it.

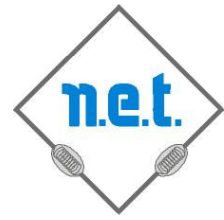
To prevent this, the NT-H2S-LI utilizes an air vent capability. This maintains equilibrium between internal and external pressures and allows the sensor to be used in any orientation and under high temperature and humidity conditions.



PERFORMANCE DATA

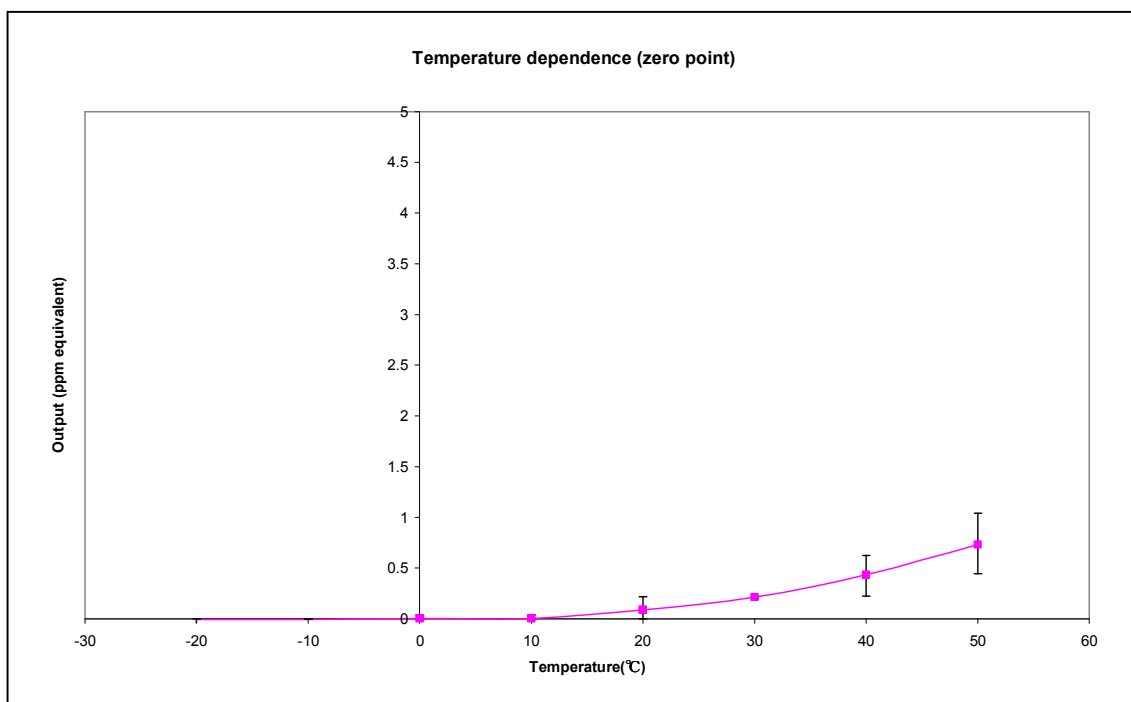
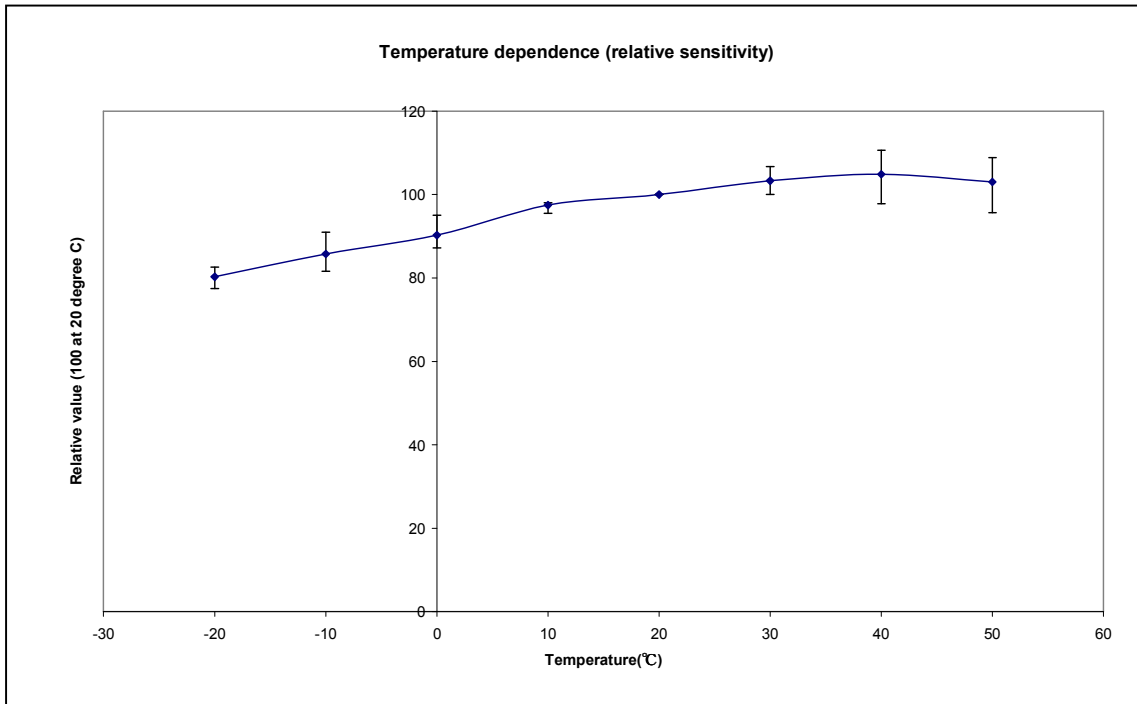
1) Linearity





2) Uncompensated Temperature Dependence:

The following graph illustrates the typical effect of temperature on the output signal of the NT-H2S-LI, for a sensor calibrated at 20°C





3) Cross Sensitivities:

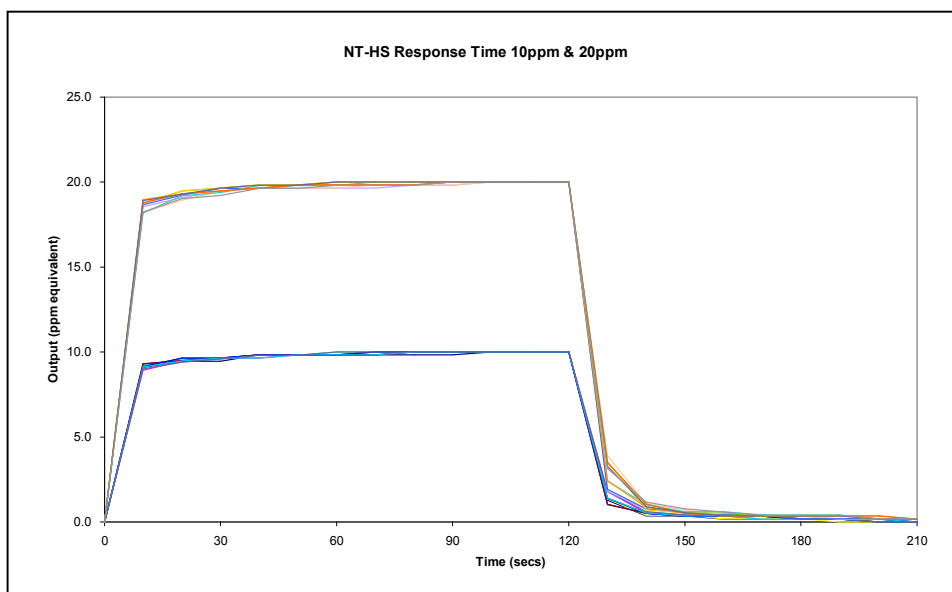
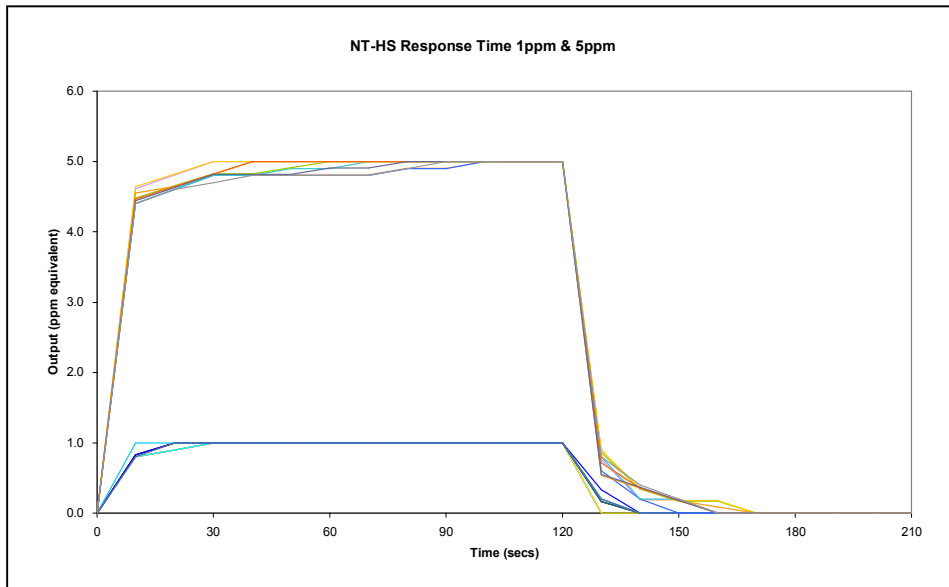
The following table gives cross sensitivity information for a variety of commonly encountered gases.

| Test Gas | Tested Concentration (ppm) | H ₂ S equivalent reading (ppm) | % cross-sensitivity |
|-------------------|----------------------------|---|---------------------|
| Carbon monoxide | 200 | <2 | <1 |
| Hydrogen Sulphide | 10 | 10 | 100 |
| Hydrogen | 1000 | <15 | <1.5 |
| Methane | 5000 | 0 | 0 |
| Carbon dioxide | 5000 | 0 | 0 |
| Sulfur dioxide | 30 | <4 | <14 |
| Nitric oxide | 10 | -0.2 to 1 | <10 |
| Nitrogen dioxide | 10 | -1.4 | 14 |
| Ammonia | 100 | 0 | 0 |
| Ethanol | 100 | 0.2 | 0.02 |
| Ethylene | 1000 | <1.2 | <0.15 |
| Chlorine | 10 | <0.5 | <5 |



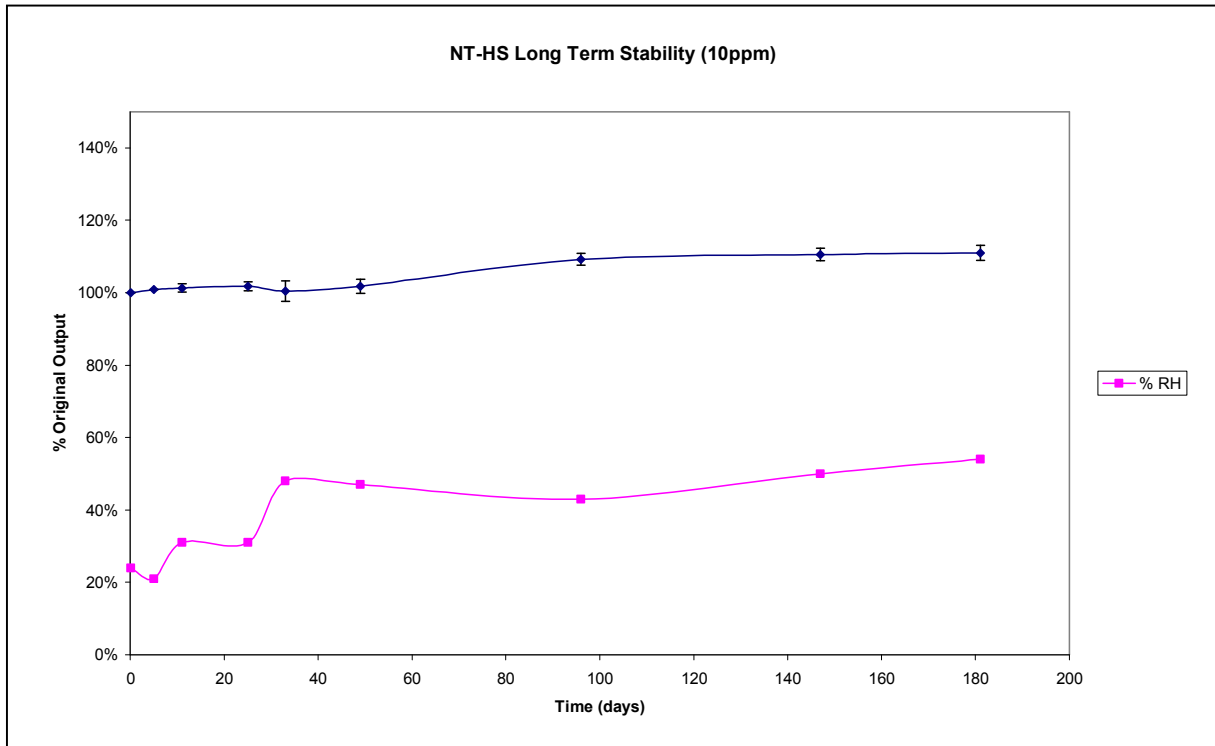
4) Response Characteristics

The following are plots of responses and recovery times for exposures to carbon monoxide at various concentrations:





5) Long-Term Drift Characteristics





DEFINITIONS

Base-line / Base-line shift

Base line means the output level in clean air. The output current value at 20°C would be less than 600nA, but this tends to increase as the ambient temperature rises more than 30°C. The base-line shift means this variation of the output level, i.e. a maximum of 600nA would be put out at 50°C. This base-line shift should be taken into account to optimise overall accuracy. In this manual, the output values are calculated to be equivalent to H₂S gas concentrations.

Gas sensitivity / Output signals

Using the NT-H2S-LI, 500 ± 100nA is generated at 1ppm of H₂S gas. For instance, the generated current value will be about 5µA at 10ppm of H₂S gas (10ppm x 500nA). This generated current is generally recorded as a voltage produced by a Current – Voltage converting circuit. In our recommended circuit, as the conversion is done through a resistor of 10 KΩ.

Response time (t₉₀)

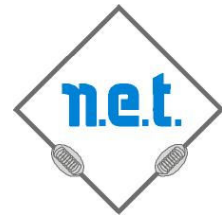
This is the time taken to reach to 90% of the maximum output value in clean air.

Repeatability

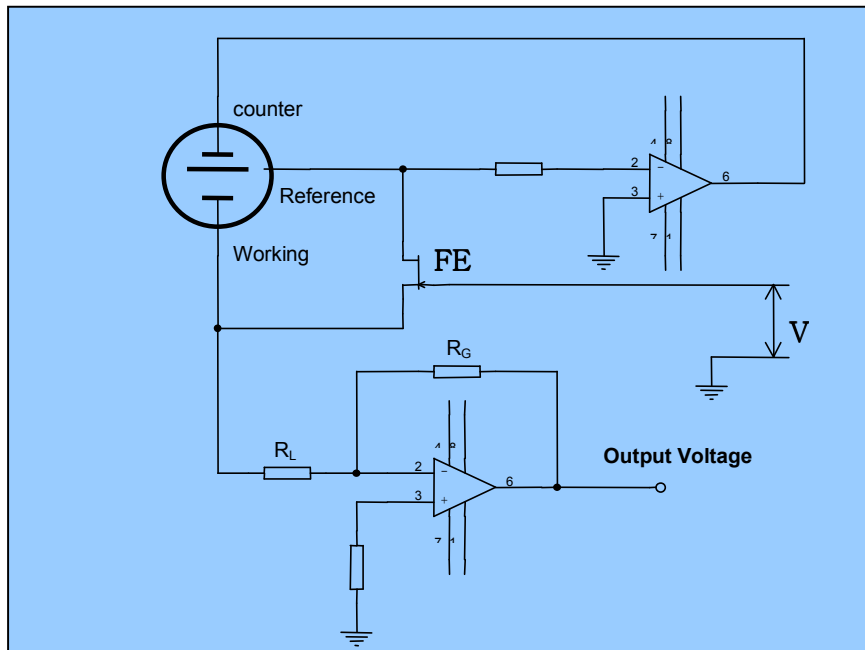
This is the maximum variation of output signals when tests are repeated under the same measuring conditions (temperature, humidity, gas concentration etc.). The repeatability of NT-H2S-LI is ± 2%, and this means that all of the test results would fall in the range of 98% ~ 102%.

Temperature dependence

All electrochemical sensors are affected by changes in the ambient temperature and the output increases as the ambient temperature rises. This is caused by the rate of oxidation reaction on the surface of the catalyst, the dispersibility of the gas in the capillary, and the thermal affects on the mobility of ions in the electrolyte. This temperature dependency can be compensated relatively easily by using a NTC thermistor.



NOTES FOR CIRCUIT DESIGN



- In this arrangement, the output voltage =

$$\text{Gas Concentration (ppm)} \times \text{Cell Output (A)} \times R_G (\Omega)$$

- So if R_G is 100k, cell output is 500nA/ppm and gas concentration is 10ppm then

$$V_{OUT} = 10 \times (500 \times 10^{-9}) \times (100 \times 10^3) = 0.5V$$

- R_L is the cell load resistor (typically 5-50 Ω). Speed of response can be increased by reducing the value of R_L , but signal noise may be increased as a consequence. The recommended values are shown on sensor datasheets
- Amplifiers are precision low input offset types, eg. OP90 or similar
- Some sensors require temperature compensation circuitry. A simple compensation network can be incorporated into this circuit by replacing R_G with a thermistor. Typically this is a NTC thermistor that has 3435K of B constant adjusting the output accuracy to within $\pm 10\%$ in the range of $-10^\circ\text{C} \sim 50^\circ\text{C}$. Any thermistor with a B-constant around 3500K and resistance value (R_{25}) of 10 K Ω can be used. Alternatively, temperature compensation may be undertaken using software lookup tables.
- For many applications, it is necessary to automatically identify critical sensor failure and in particular open and short circuit conditions. Nemoto has developed a patented electrical method suitable for battery and mains voltage operation. If this is required, please contact Nemoto and we will be happy to provide further details subject to a non-disclosure agreement.



NOTES

1) Long-term drift of gas sensitivity

All electrochemical gas sensors lose sensitivity over time due to small changes on the surface of the working electrode, reducing its oxidation capability. To reduce this, the NT-H2S-LI uses a newly developed electrode catalyst that will not deteriorate by more than 5% / year. Typically, these changes are limited to less than 5% but we recommend that this deterioration should be taken into account when designing application circuits.

2) Environmental affects on gas sensitivity

Due to the hygroscopic nature of the electrolyte used in electrochemical sensors, moisture is absorbed from or released to the surrounding atmosphere. In high humidity moisture is absorbed, causing an increased sensitivity. In low humidity moisture is released back to the atmosphere and the sensitivity decreases. Conventional electrochemical gas sensors show annual variation of gas sensitivity as much as 10 ~ 20%.

The NT-H2S-LI utilises an advanced electrolyte management design and, combined with the unique electrode catalyst structure, these changes can be greatly reduced. Under normal operating conditions, gas sensitivity should change by no more than $\pm 5\%$ of the output value. The NT-H2S-LI is designed so that all changes due to moisture uptake/release are completely reversible.

If the gas intake area of the sensor is blocked with water drops or other liquid, gas cannot enter the sensor. The NT-H2S-LI is fitted with an integral hydrophobic barrier to prevent this, but we recommend the use of additional membrane barriers if the sensor is in highly condensing RH conditions. H₂S gas is almost the same weight as air, but as it is a combustion by-product it will usually be heated and therefore rise. H₂S detecting devices should be installed at least 5ft from the floor and preferable on the upper part of walls or ceilings.

If the sensor is to be used in more irregular atmospheres, please contact us for assistance.

3) Storage of sensors

Electrochemical sensors should be stored in a clean air under room temperature, preferably 0°C ~ 20°C and in non-condensing RH conditions. The maximum storage period would be 6 months after delivery. For sensors stored for more than 6 months, the performance guaranteed period will be shortened by the excess storage period. Unlike semiconductor type or hot-wire type gas sensors, the gas sensitivity of electrochemical gas sensors will change as time passes regardless of whether sensor has been used or not.



4) Mounting sensors

Electrode pins must be connected correctly to ensure operation.

A thermistor for temperature compensation must be located near the sensor and away from heat sources such as transformer.

NT-H2S-LI can be mounted in any orientation.

NT-H2S-LI pins cannot be soldered as excess heating may cause the deformation of the housing and eventually leakage of electrolyte.

5) Calibration and gas testing

Calibration of detectors or densitometers should be done after the output value has been stabilized in clean air.

Evaluation of gas sensitivity should be made in clean, humidified, air. When a test gas is blown directly to the gas intake area, higher gas sensitivity may be observed. It is therefore best to test and calibrate gas detection instruments and sensors in diffusion mode. This can be achieved by using a suitable test housing where a low flow rate is used (<1l/min) and where the air is agitated to ensure equal gas diffusion throughout.

6) Other

Unless otherwise advised by Nemoto, voltage should not be supplied directly to the electrode pins.

Do not bend the pins.

Do not apply more than 5 kgs/cm² to the sensor.

Take care not to block the gas intake area as it may prevent gas entering the sensor.

Never put foreign material in the gas intake area as it may cause the electrolyte leakage

Do not expose the sensor to excess vibration or shock.

If the sensor housing is damaged, do not use the sensor.

After the sensor is exposed to high concentration gas for long periods, the output signal may require time to recover to normal operation.

Do not blow organic solvents, paints, chemical agents, oils, or high concentration gases directly onto sensors.

Do not disassemble the sensor as this may cause electrolyte leakage.